



ece|OC

Task Force
Sufficiency of aquatic hazard
information for environmental
risk assessment

Topical Scientific Workshop on Soil Risk Assessment
7-8 October 2015, ECHA, Helsinki



Chemours™

Dr. Michiel Claessens
michiel.claessens@chemours.com

1. Background
2. General methods - Data collection and cleanup
3. Methods, Results & Discussion
 - a) Hazard comparison (I) - Direct comparison
 - b) Hazard comparison (II) - Role of phys-chem & MOA
 - c) Hazard comparison (III) - Chemical activity
 - d) How significant is all of this in a risk assessment? - PNEC comparison
4. Conclusions & Suggestions

- REACH ecotoxicological data requirements

- 1 to 10 t/yr: only acute aquatic data
- 10 to 100 t/yr: only acute aquatic data
- 100 to 1000 t/yr: chronic aquatic and acute terrestrial data
=> EqP can be used if no data available
- > 1000 t/yr: chronic terrestrial included

→ Environmental hazard and risk assessments under REACH depend strongly on aquatic ecotoxicity data

Background

- REACH aquatic hazard and risk assessment

- What's the hazard for aquatic species?

→ $PNEC_{\text{aquatic}} = 0,025 \text{ mg/L}$



EC50 (mg/L)

25

32

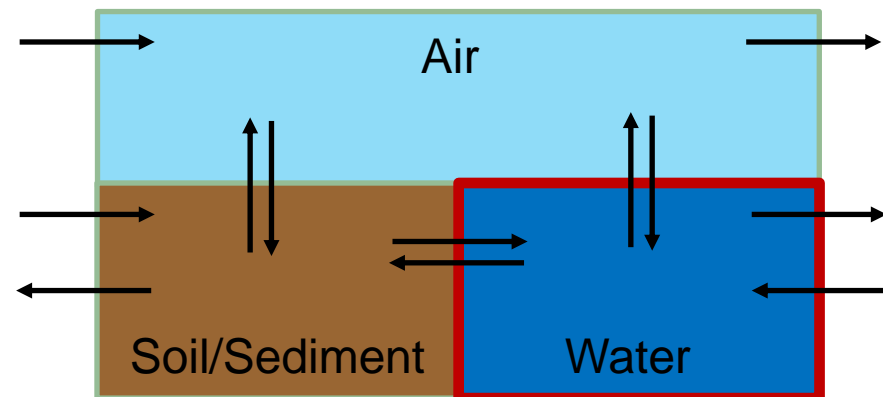
> 100

$$\frac{25 \text{ mg/L}}{AF} = \frac{25 \text{ mg/L}}{1000} = 0.025 \text{ mg/L}$$

- To what levels are aquatic species exposed?

→ $PEC_{\text{aquatic}} = 0,005 \text{ mg/L}$

$$RCR = \frac{PEC_{\text{aquatic}}}{PNEC_{\text{aquatic}}} = \frac{0.005}{0.025} = 0.2$$



Background

■ REACH soil hazard and risk assessment - option A

- What's the hazard for soil species?

$$\rightarrow \text{PNEC}_{\text{soil,AF}} = 0,264 \text{ mg/kg}$$



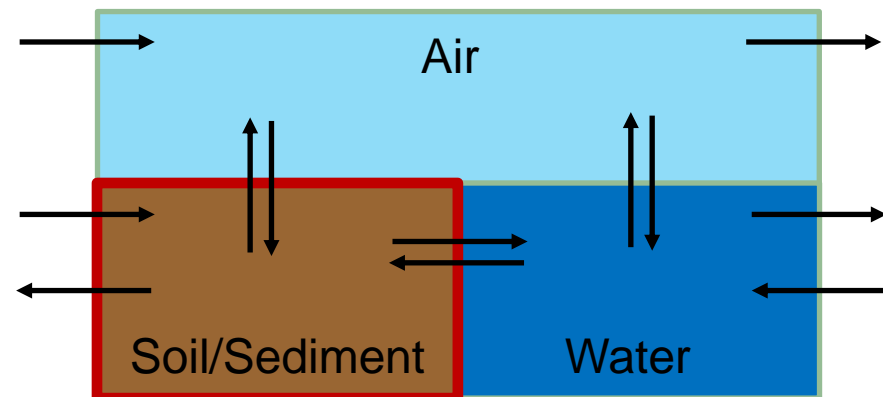
EC50 (mg/kg) 264

$$\frac{264 \text{ mg/kg}}{\text{AF}} = \frac{264 \text{ mg/kg}}{1000} = 0.264 \text{ mg/kg}$$

- To what levels are soil species exposed?

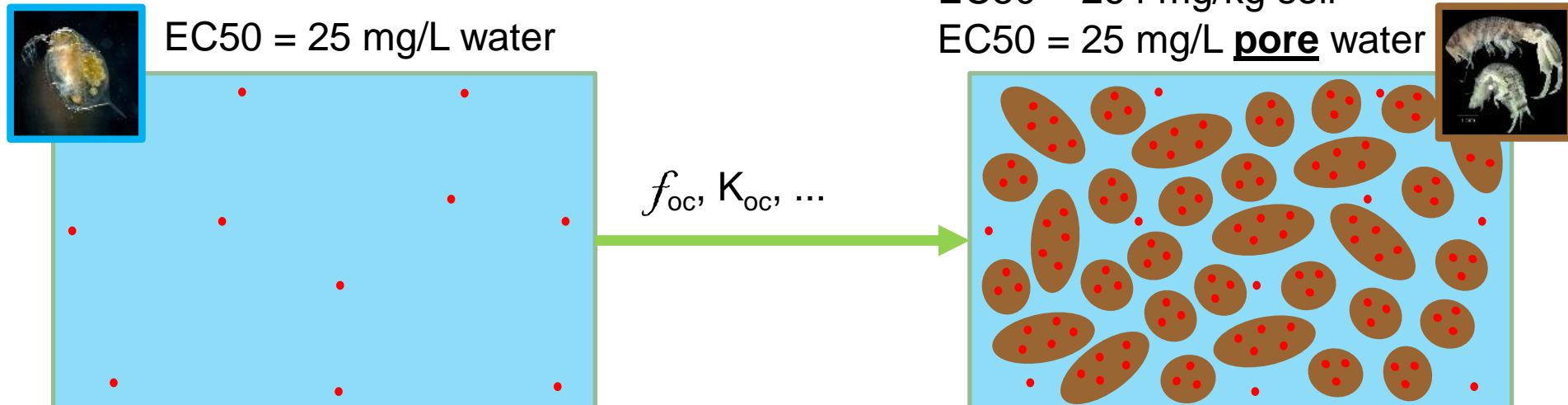
$$\rightarrow \text{PEC}_{\text{soil}} = 0,05 \text{ mg/kg}$$

$$\text{RCR} = \frac{\text{PEC}_{\text{soil}}}{\text{PNEC}_{\text{soil,AF}}} = \frac{0.05}{0.264} = 0.2$$



Background

- REACH soil hazard and risk assessment - option B
 - What's the hazard for soil species?
 - $PNEC_{soil}$ derived from $PNEC_{aquatic}$ using equilibrium partitioning (**EqP**)
 - Aquatic and soil organisms are equally sensitive
 - Toxicant uptake via water phase



REACH soil PNEC derivation

Option A

- Experimental soil EC50
- Assessment factor (AF)



$PNEC_{soil,AF}$

Option B

- Experimental aquatic EC50
- Equilibrium partitioning (EqP)



$PNEC_{soil,EqP}$

?

=

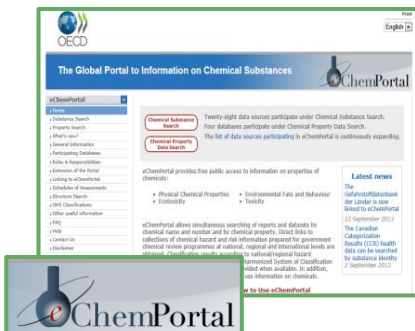
Do we miss any risks when using $PNEC_{soil,EqP}$?

Terms of reference (excerpt)

- Confirm the **extent** to which the existing system is describing the **aquatic hazard** of substances and is **protective** for risks in the compartment (water, soil, and sediment).
- Identify those substances for which a **better understanding** is required.
- Review available information acquisition strategies for obtaining information in soil and sediment with the aim of **identifying reliable approaches** for improving the confidence in the environmental safety assessment.

Exercise with a regulatory focus!

Data mining



eChem Portal

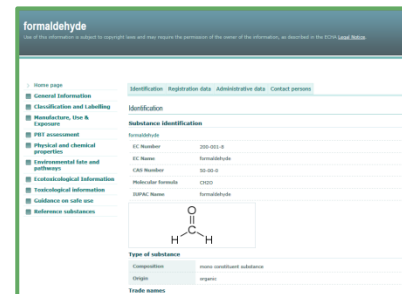
Develop a database containing substances with:

- Aquatic and terrestrial (soil) toxicity data
- (Aquatic and sediment toxicity data)

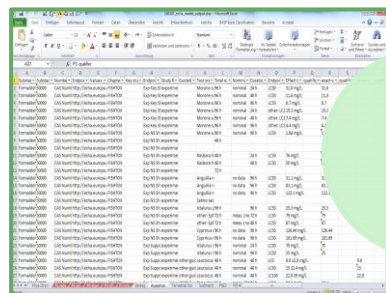
~ **500 substances** have been identified carrying both aquatic and terrestrial data (Klimisch score 1 & 2 only).

Additional data extracted from ECHA webpage: **Phys-Chem, biodegradation, PNECs**

ECHA website



xlsx



The list covers **12 182 endpoint study records.**

Data mining



- Ecotoxicity data, acute and chronic
 - Aquatic: algae (& other aquatic plants), invertebrates and fish; PNECs
 - Terrestrial plants, soil dwelling organisms, soil microorganisms; PNECs
 - Phys-Chem data: Kow, Koc, Vapour pressure, Henry's law constant, water solubility
- Data cleanup: units, endpoints, exposure duration, ...

■ Additions

- Chemical class, molecular weight and SMILES code as derived by ECOSAR
- Mode of action: Modified VERHAAR class (via ToxTree)
- QSAR phys-chem parameters => data gap filling (EPIsuite)

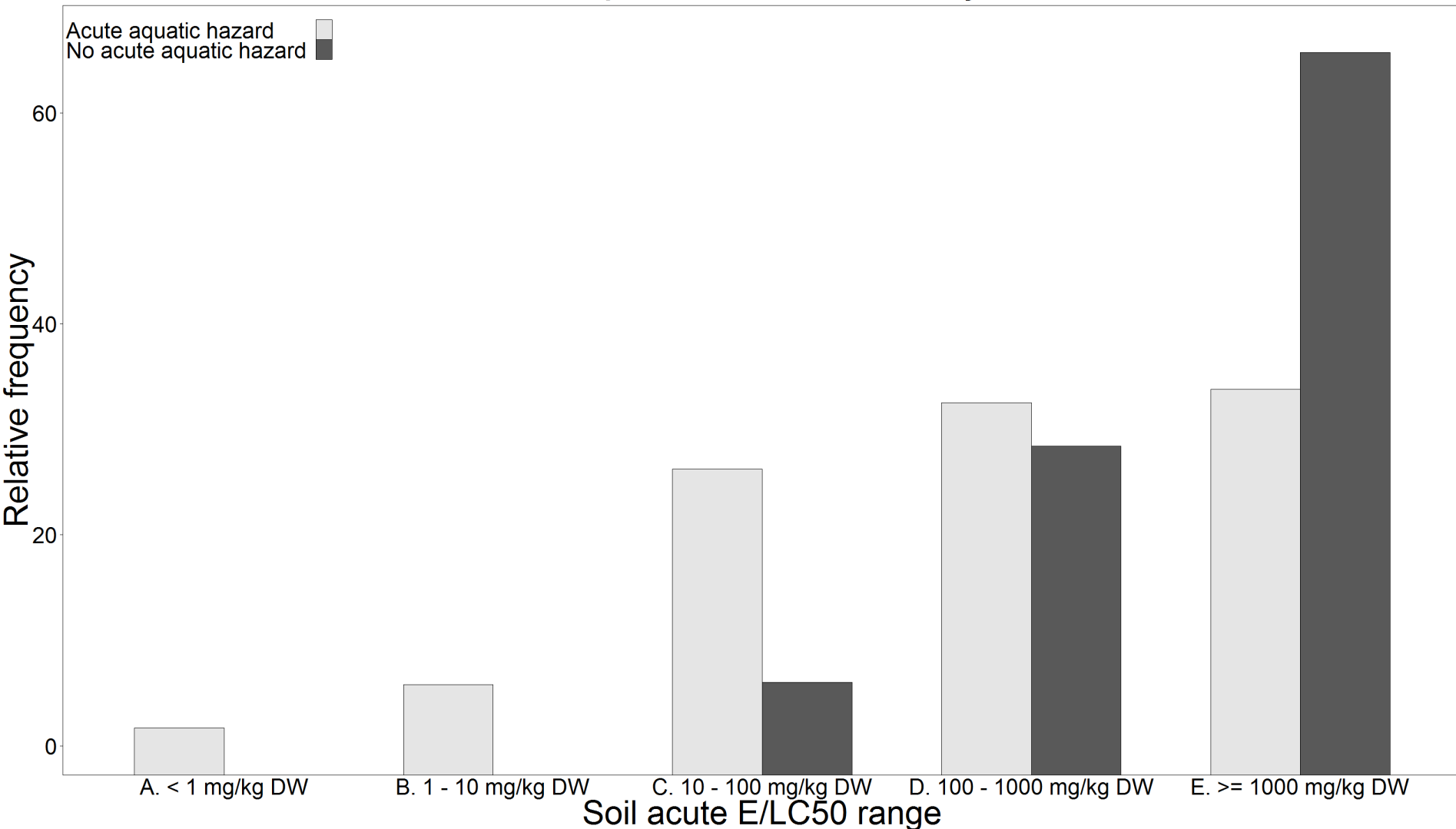
- Unit conversions (e.g. ng/L, g/L, mol/L, ... => mg/L)

| | A | B | C | D | E | G | K | AG | AM | AN | | |
|----|----------------|------------------|------------------|------------------|----------------|--------------------------|--------------|---------------------|----------------------------|----------|----------|------------------------|
| 1 | Substance Name | Substance Number | Number type | SMILES | VERHAAR Values | Study Result Type | Value | Test type | Partition coefficient type | | | |
| 2 | formaldehyde | 50000 CAS Number | O=C | | | estimated by calculation | 0.35 log Pow | octanol-water | | | | |
| 3 | formaldehyde | 50000 CAS Number | O=C | | | estimated by calculation | 0.35 log Pow | octanol-water | | | | |
| 4 | formaldehyde | 50000 CAS Number | O=C | | | experimental result | 0.35 log Pow | octanol-water | | | | |
| 5 | form | | | | | | | | | | | |
| 6 | form | 1 | Substance Name | Substance Number | Number type | SMILES | VERHAAR | Study Result Type | Test organism | Duration | Endpoint | Effect concentration |
| 7 | form | 2 | formaldehyde | 50000 CAS Number | O=C | | 3 | experimental result | Morone saxatilis | 24 h | LC50 | 31.8 mg/L |
| 8 | form | 3 | formaldehyde | 50000 CAS Number | O=C | | 3 | experimental result | Morone saxatilis | 48 h | LC50 | 11.8 mg/L |
| 9 | form | 4 | formaldehyde | 50000 CAS Number | O=C | | 3 | experimental result | Morone saxatilis | 96 h | LC50 | 6.7 mg/L |
| 10 | form | | | | | | | | | | | |
| 11 | form | 1 | Substance Name | Substance Number | Number type | SMILES | VERHAAR | Study Result Type | Test organism | Duration | Endpoint | Effect conc (ECHA) |
| 12 | form | 2 | formaldehyde | 50000 CAS Number | O=C | | 3 | experimental result | Eisenia fetida | 48 h | LC50 | 1 other: 10 g/cm2 |
| 13 | form | 3 | formaldehyde | 51036 CAS Number | O=C(O)O | | 1 | experimental result | Eisenia fetida | 56 d | NOEC | 30 mg/kg soil dw |
| 14 | form | 4 | formaldehyde | 51036 CAS Number | O=C(O)O | | 1 | experimental result | Eisenia fetida | 56 d | LC50 | 65.8 mg/kg soil dw |
| 15 | form | | | | | | | | | | | |
| 16 | form | 9 | 5-[[2 | | | | | AT | AV | AW | AX | |
| 17 | form | 10 | formaldehyde | 50000 CAS Number | O=C | | 3 | STP | partition coefficient | | PNEC AF | PNEC value |
| 18 | form | 11 | formaldehyde | 50000 CAS Number | O=C | | 3 | Freshwater | partition coefficient | | | 10 0.47 mg/L |
| 19 | form | 12 | formaldehyde | 50000 CAS Number | O=C | | 3 | Freshwater | partition coefficient | | | 10 0.47 mg/L |
| 20 | form | 13 | formaldehyde | 50000 CAS Number | O=C | | 3 | Freshwater sediment | assessment factor | | | 2.44 mg/kg sediment dw |
| 21 | form | 14 | formaldehyde | 50000 CAS Number | O=C | | 3 | Marine water | assessment factor | | | 10 0.47 mg/L |
| 22 | form | 15 | formaldehyde | 50000 CAS Number | O=C | | 3 | Marine sediment | partition coefficient | | | 2.44 mg/kg sediment dw |
| 23 | form | 16 | formaldehyde | 50000 CAS Number | O=C | | 3 | Soil | partition coefficient | | | 0.21 mg/kg soil dw |
| 24 | form | 17 | amino(imino)meth | 50011 CAS Number | NC(N)=N | | | STP | assessment factor | | | 10 0.29 mg/L |
| 25 | form | 18 | amino(imino)meth | 50011 CAS Number | NC(N)=N | | | Freshwater | assessment factor | | | 10 0.29 mg/L |

Hazard comparison (I) – Direct comparison



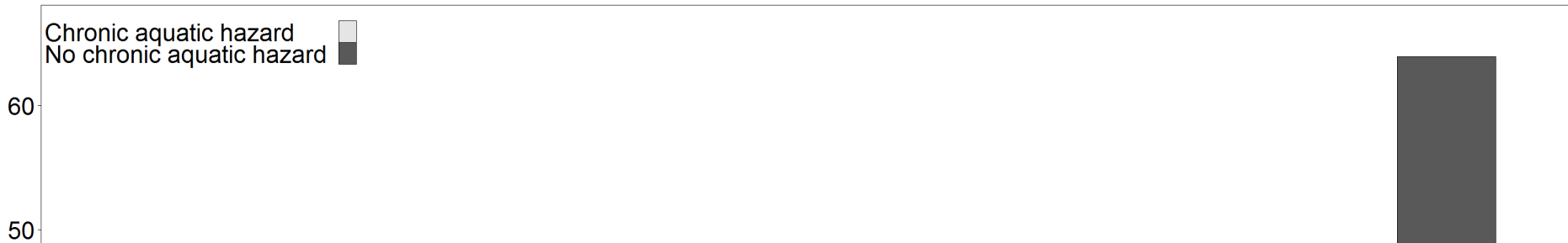
Acute aquatic hazard vs. soil toxicity



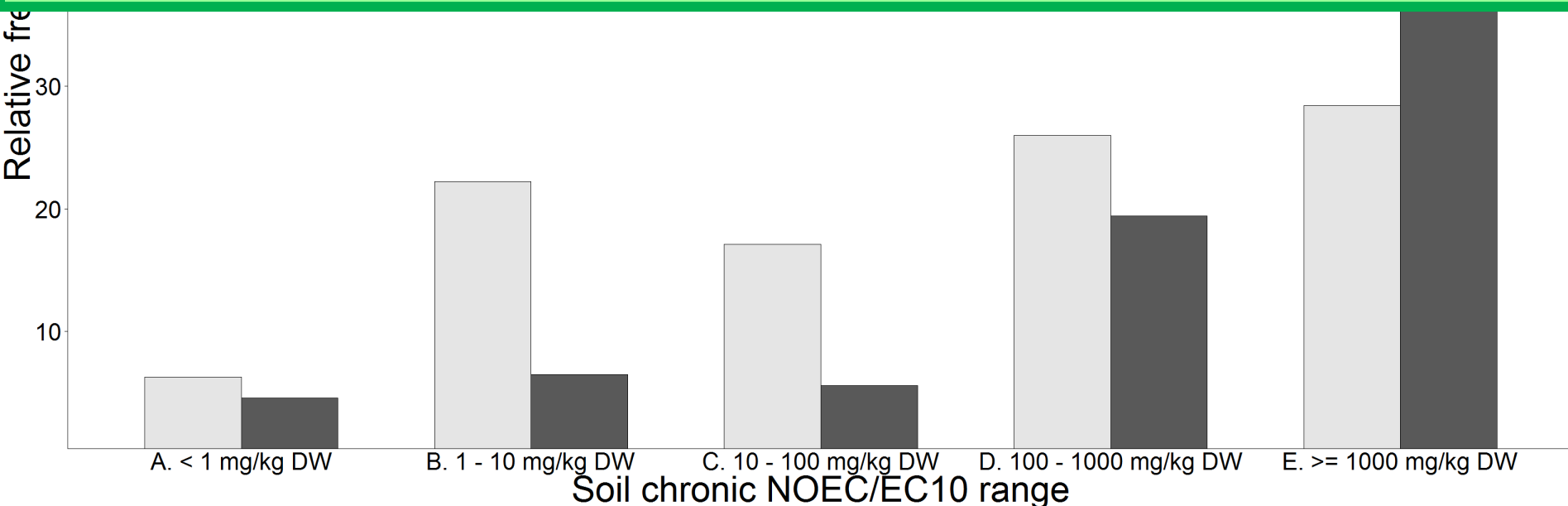
Hazard comparison (I) – Direct comparison



Chronic aquatic hazard vs. soil toxicity



If there is no aquatic hazard identified, ~ 60% of soil ecotoxicity data > 1000 mg/kg



Hazard comparison (II) – Role of phys-chem & MOA



- Unit conversion
 - mg/kg dry (or wet) weight => mg/L pore water
 - Equilibrium partitioning method (EPM; formulas from ECHA guidance)
- Why EPM?
 - Comparability of units
 - No hazard threshold for soil
 - The concept makes sense and offers a good basis for this exercise
=> in which cases does the hazard to soil and sediment organisms deviate from what EP theory would predict?

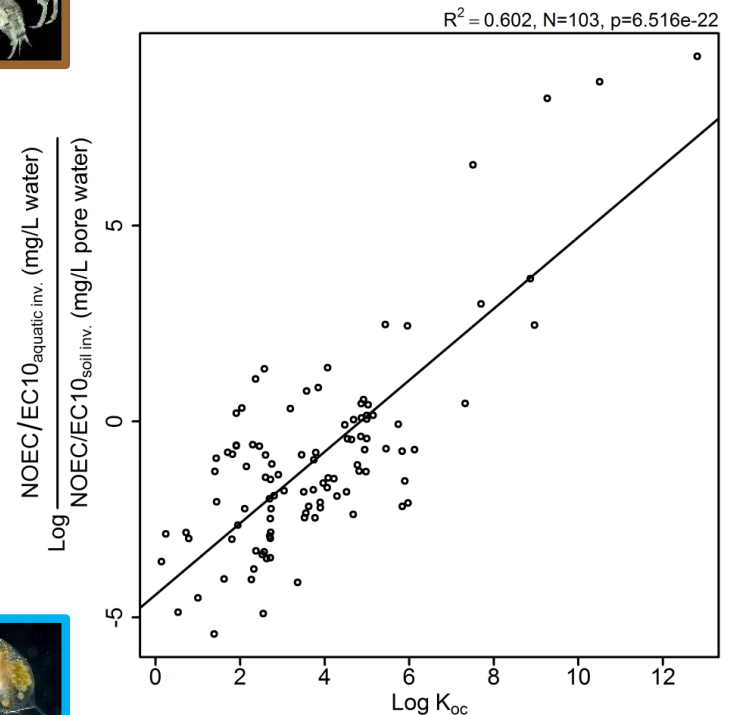
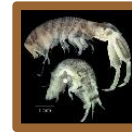
Hazard comparison (II) – Role of phys-chem & MOA

- Data presentation

$$\text{Log} \frac{\text{EC50}_{\text{aqua,inv}}}{\text{EC50}_{\text{soil,inv}}}$$

$> 0 \rightarrow \text{EC50}_{\text{aqua,inv}} > \text{EC50}_{\text{soil,inv}}$
 \rightarrow Soil invertebrates are more sensitive

$< 0 \rightarrow \text{EC50}_{\text{aqua,inv}} < \text{EC50}_{\text{soil,inv}}$
 \rightarrow Soil invertebrates are less sensitive

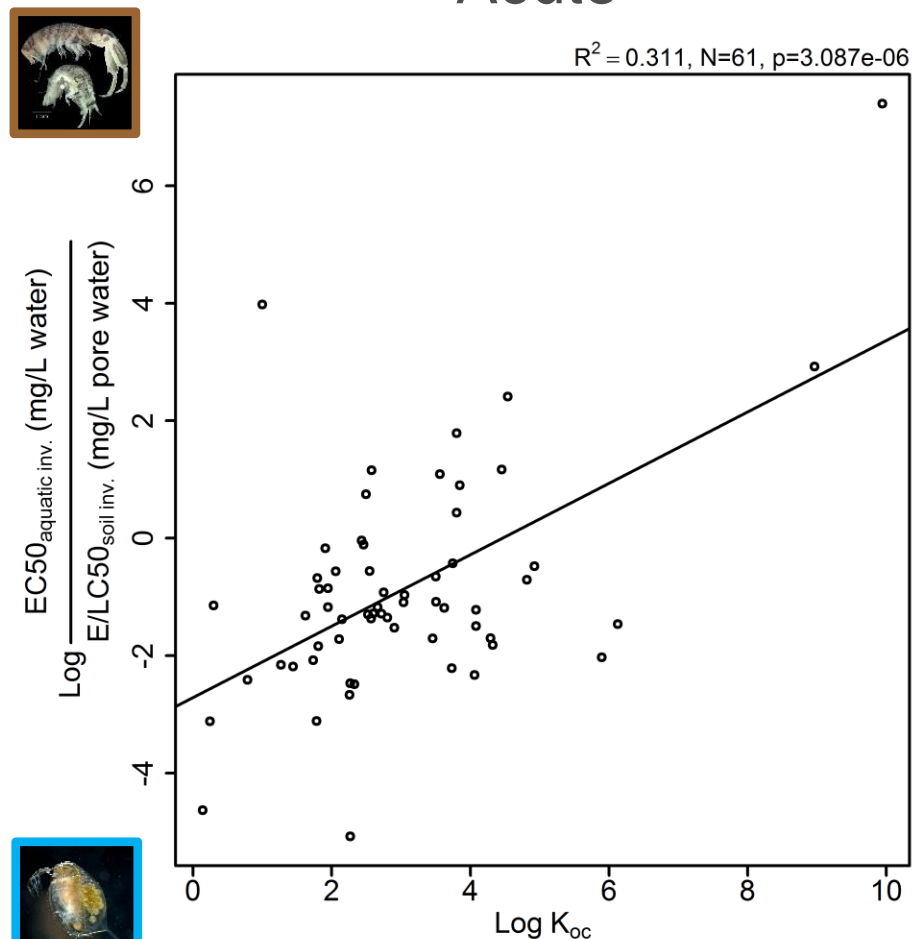


Hazard comparison (II) – Role of phys-chem & MOA

PC - Versus Log K_{oc} (aquatic vs soil invertebrates)

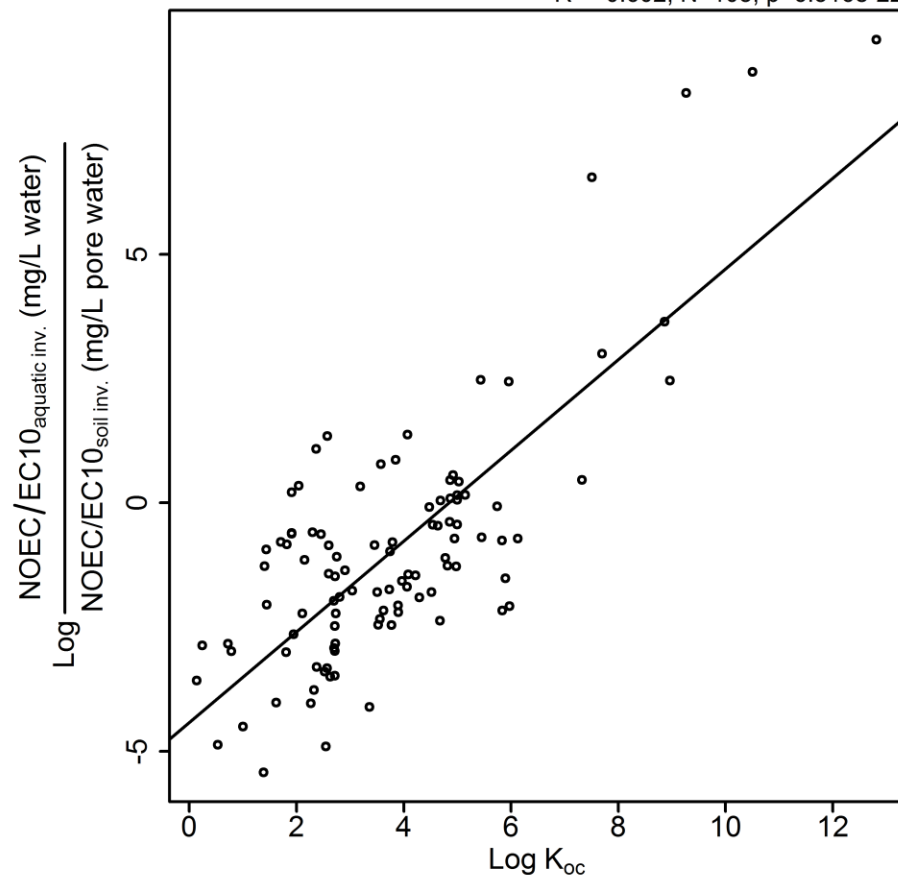
Acute

$R^2 = 0.311, N=61, p=3.087e-06$



Chronic

$R^2 = 0.602, N=103, p=6.516e-22$

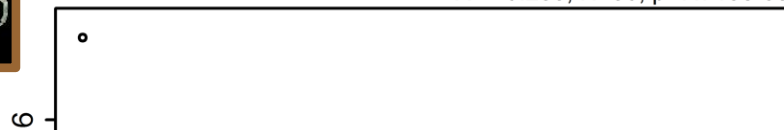


Hazard comparison (II) – Role of phys-chem & MOA

PC - Versus Log VP (aquatic vs soil invertebrates)

Acute

$R^2 = 0.253$, $N=56$, $p=7.715e-05$



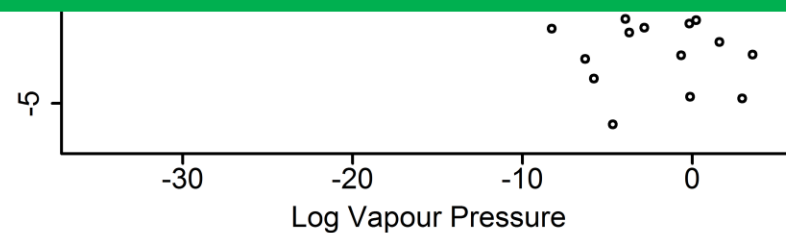
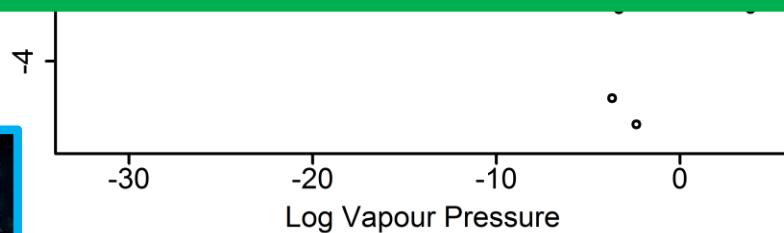
Chronic

$R^2 = 0.252$, $N=92$, $p=3.354e-07$



- **The hazard to soil invertebrates relative to aquatic organisms, seems to increase with rising K_{oc} (and declining VP?)**

- Due to the rising importance of contaminant uptake via food?
- Experimental artifact in the aquatic and/or soil ecotoxicity studies?
 - Declining concentrations?
 - Non-equilibrium?



Hazard comparison (II) – Role of phys-chem & MOA

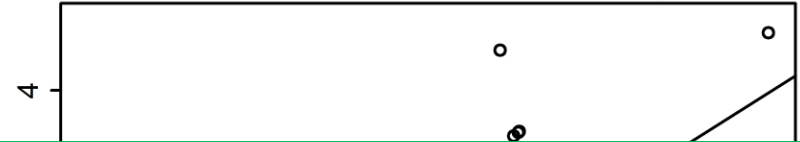
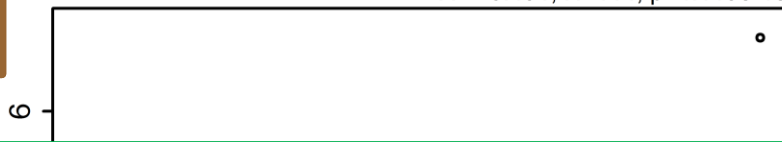
PC - Versus Log K_{oc} (Algae vs terrestrial plants)

Acute

$R^2 = 0.491$, $N=114$, $p=4.119e-18$

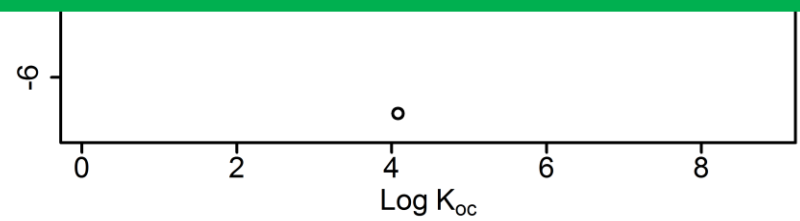
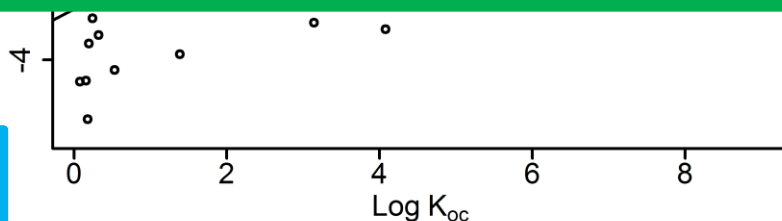
Chronic

$R^2 = 0.401$, $N=77$, $p=6.215e-10$



- **The hazard to soil invertebrates relative to aquatic organisms, seems to increase with rising K_{oc} (and declining VP?)**

- Due to the rising importance of contaminant uptake via food?
- Experimental artifact in the aquatic and/or soil ecotoxicity studies?
 - Declining concentrations?
 - Non-equilibrium?

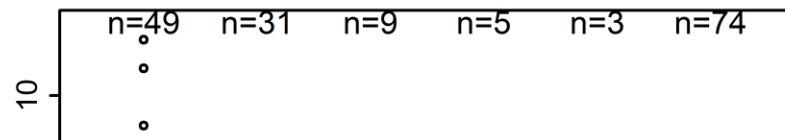
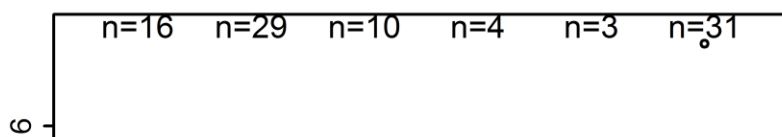


Hazard comparison (II) – Role of phys-chem & MOA

MOA - Versus VERHAAR (aquatic vs soil invertebrates)

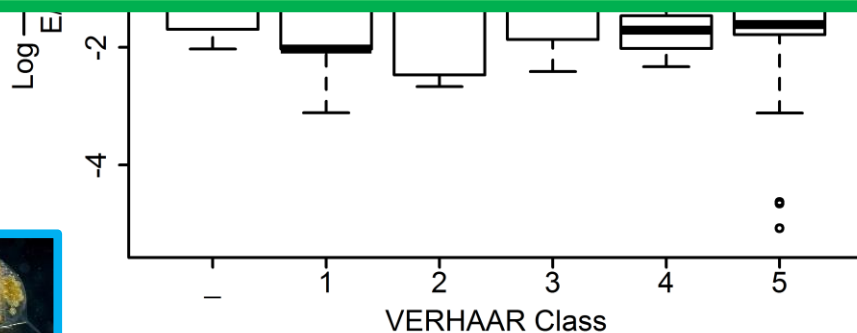
Acute

Chronic



• Based on VERHAAR class, there is no clear difference in sensitivity between aquatic and soil species for any of the MOAs, but keep in mind that:

- Only industrial chemicals included
- VERHAAR class may not be detailed enough

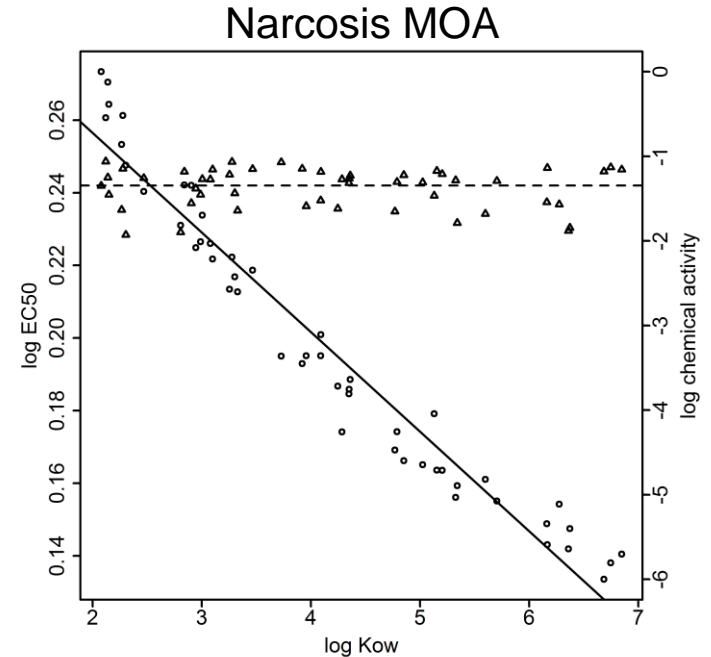


Hazard comparison (III) – Chemical activity

Chemical activity is the fraction of saturation

$$\gamma = \frac{\text{Concentration}}{\text{Solubility}} \not> 1$$

Oversaturation ($\gamma > 1$):



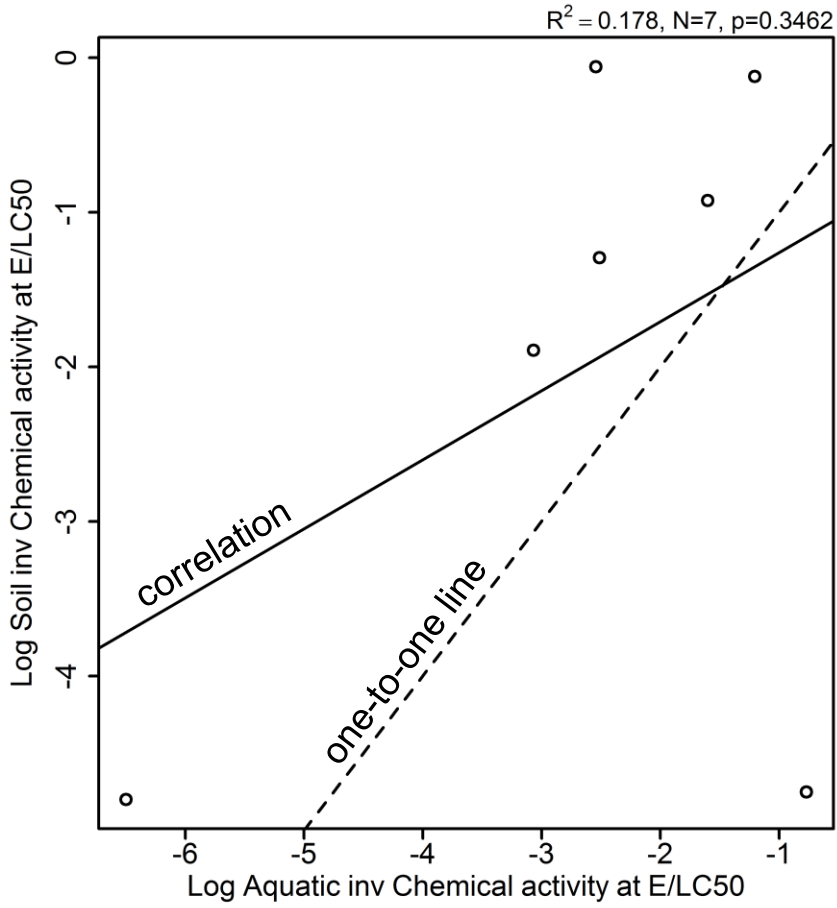
| | Acute | Chronic |
|---------|-------|---------|
| Aquatic | 37.5% | 52.9% |
| Soil | 84.5% | 74.1% |

Hazard comparison (III) – Chemical activity

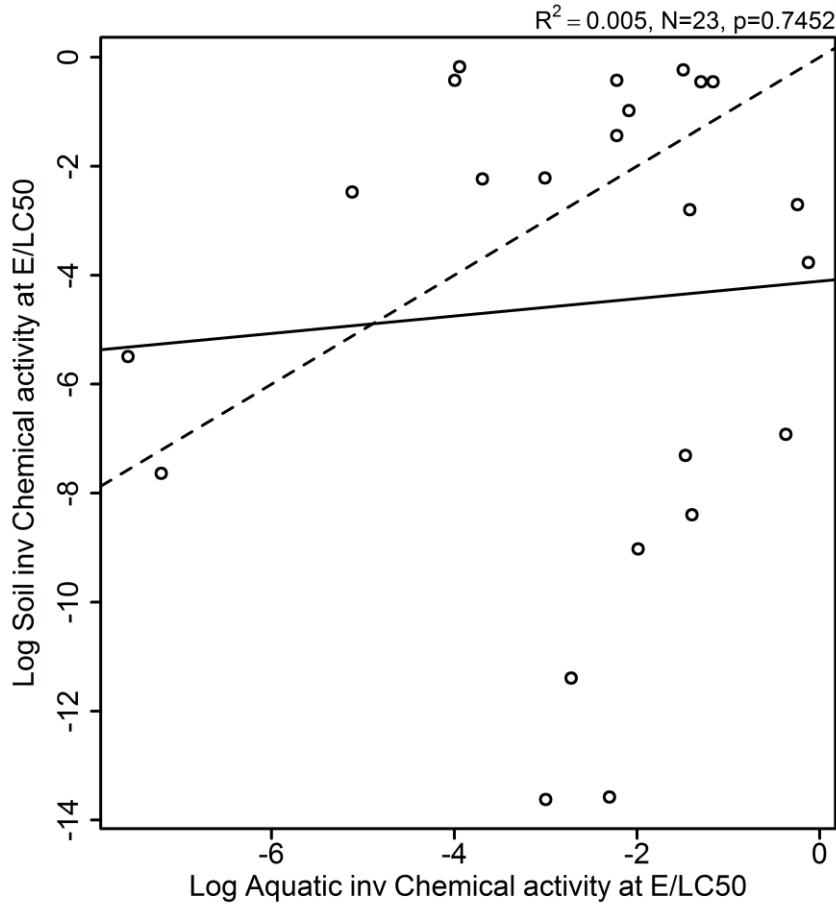


Aquatic vs. Soil invertebrates

Acute



Chronic



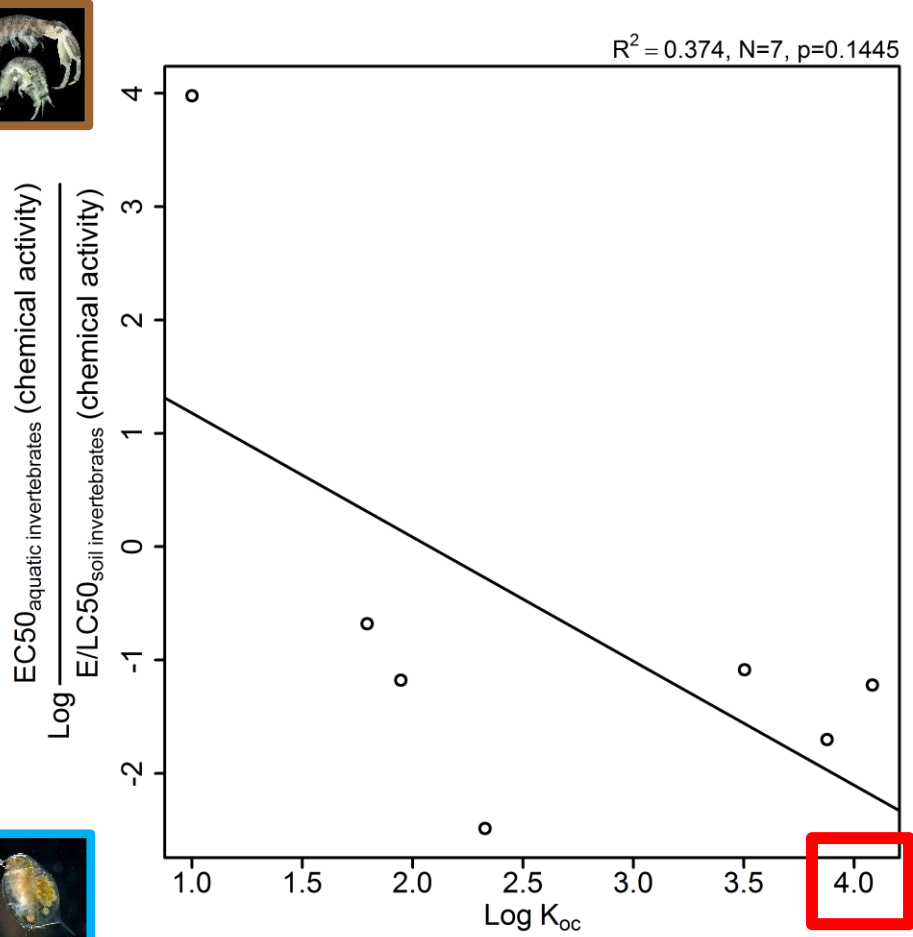
Hazard comparison (III) – Chemical activity



PC – Versus log Koc

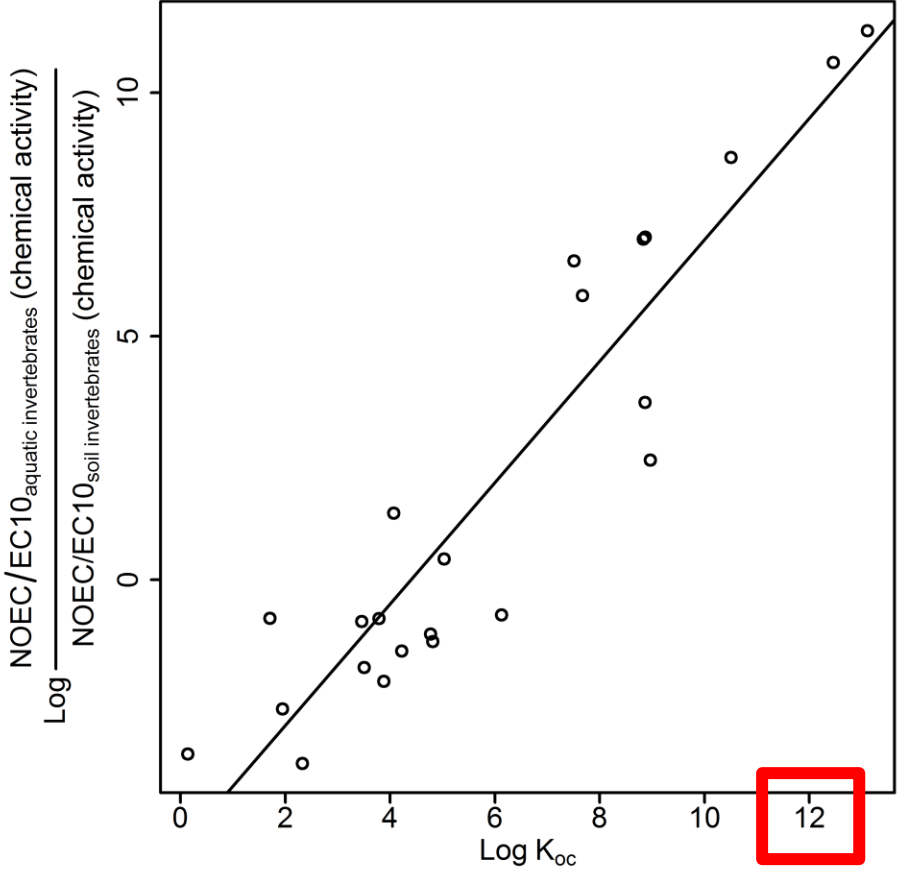
Acute

$R^2 = 0.374, N=7, p=0.1445$



Chronic

$R^2 = 0.868, N=23, p=1.058e-10$



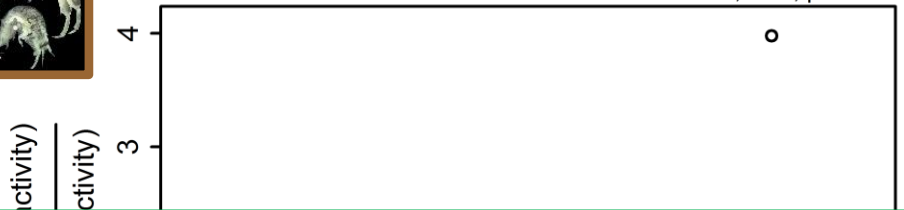
Hazard comparison (III) – Chemical activity



PC – Versus Vapour pressure

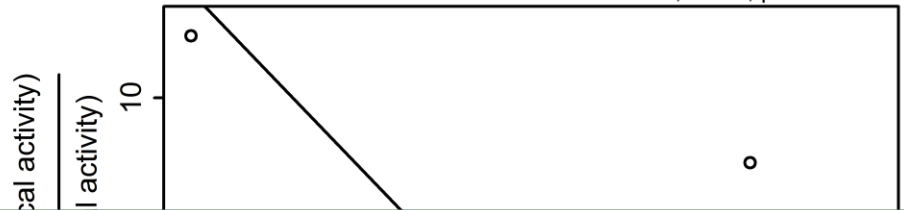
Acute

$R^2 = 0.02, N=7, p=0.762$

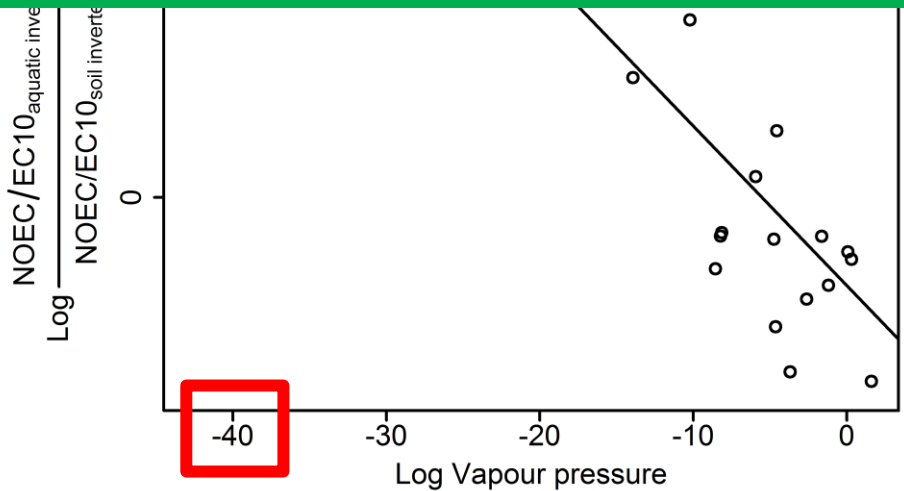
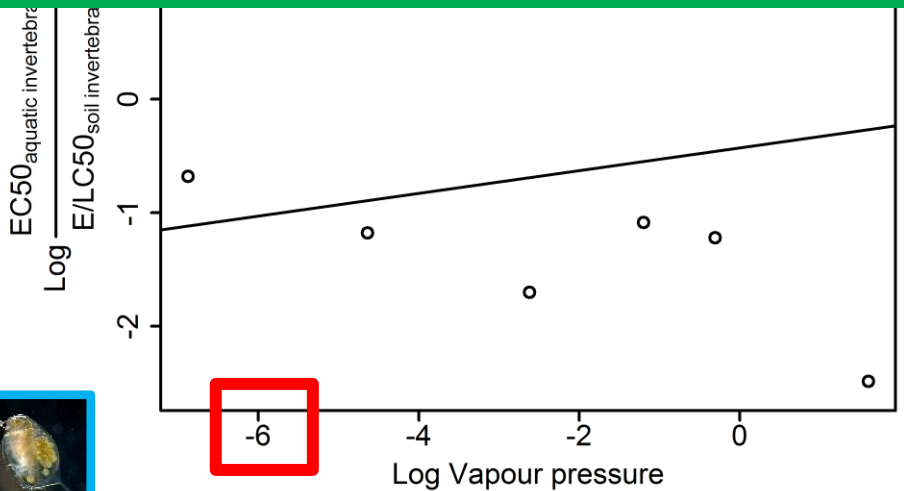


Chronic

$R^2 = 0.633, N=19, p=4.611e-05$



Again, this seems to point to a difference in partitioning or state of equilibrium



REACH soil PNEC derivation

Option A

- Experimental soil EC50
- Assessment factor (AF)



$PNEC_{soil,AF}$

Option B

- Experimental aquatic EC50
- Equilibrium partitioning (EqP)



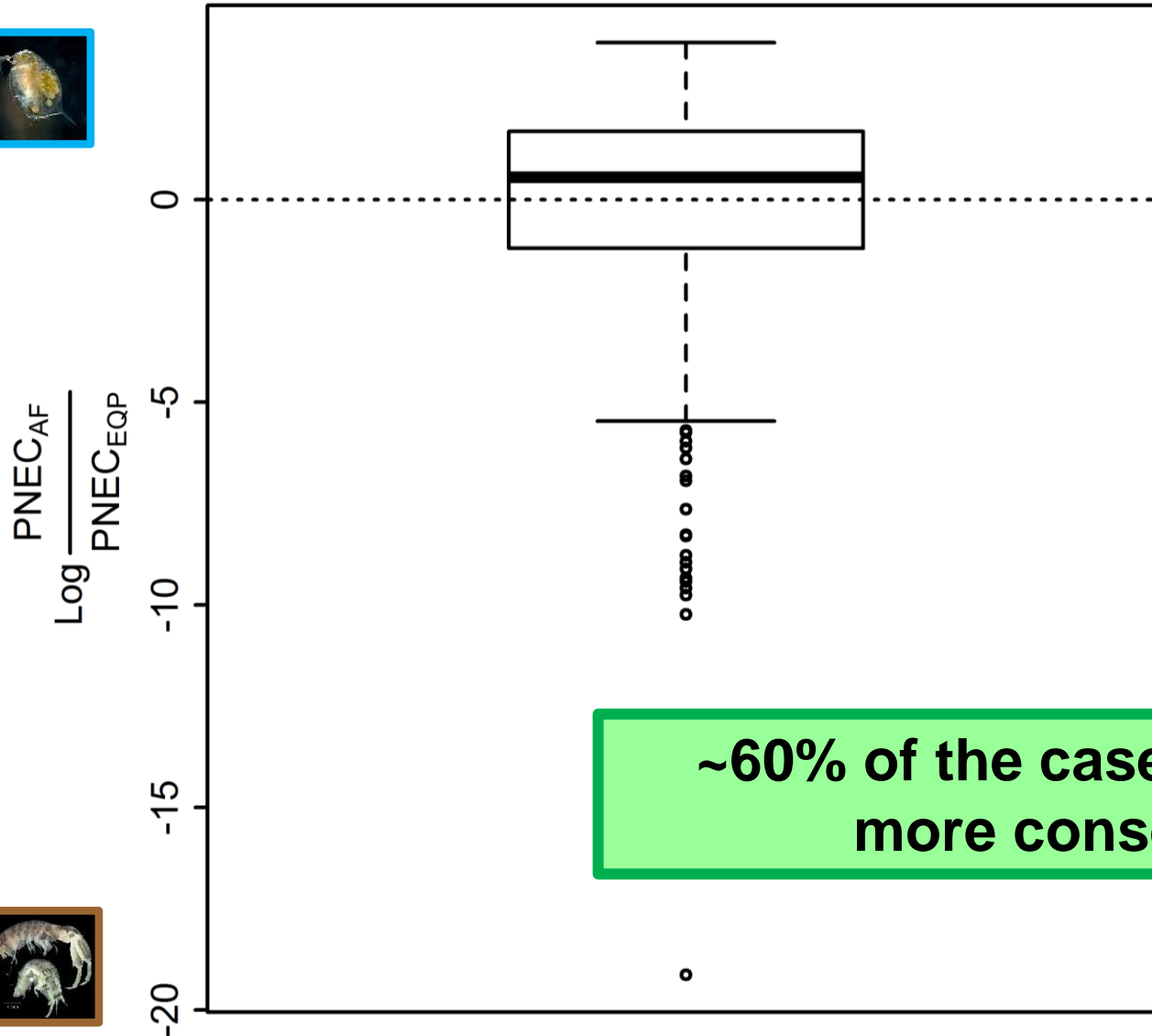
$PNEC_{soil,EqP}$

?

=

Do we miss any risks when using $PNEC_{soil,EqP}$?

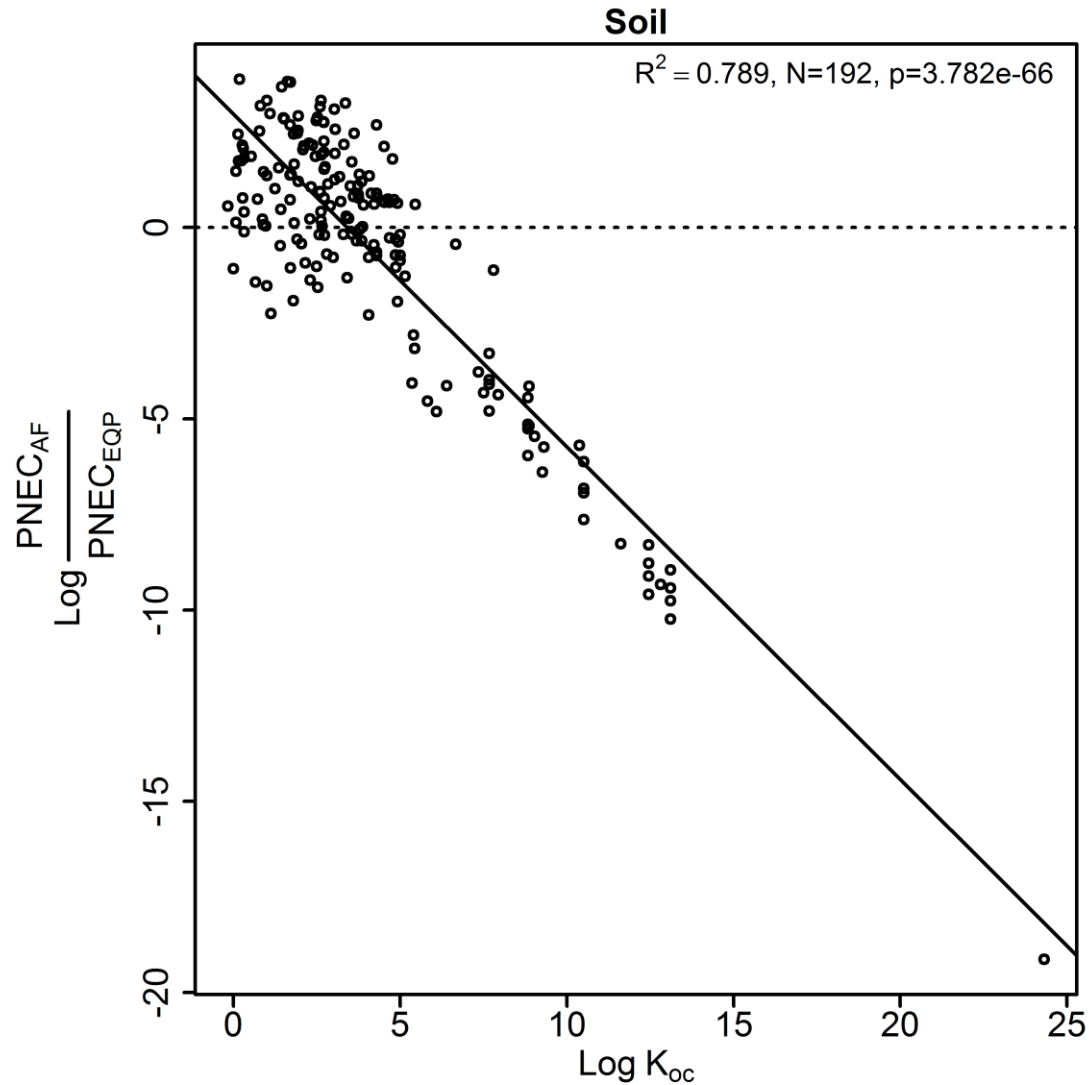
Risk assessment (IV) – PNEC/Risk



~60% of the cases PNEC_{EQP} is more conservative



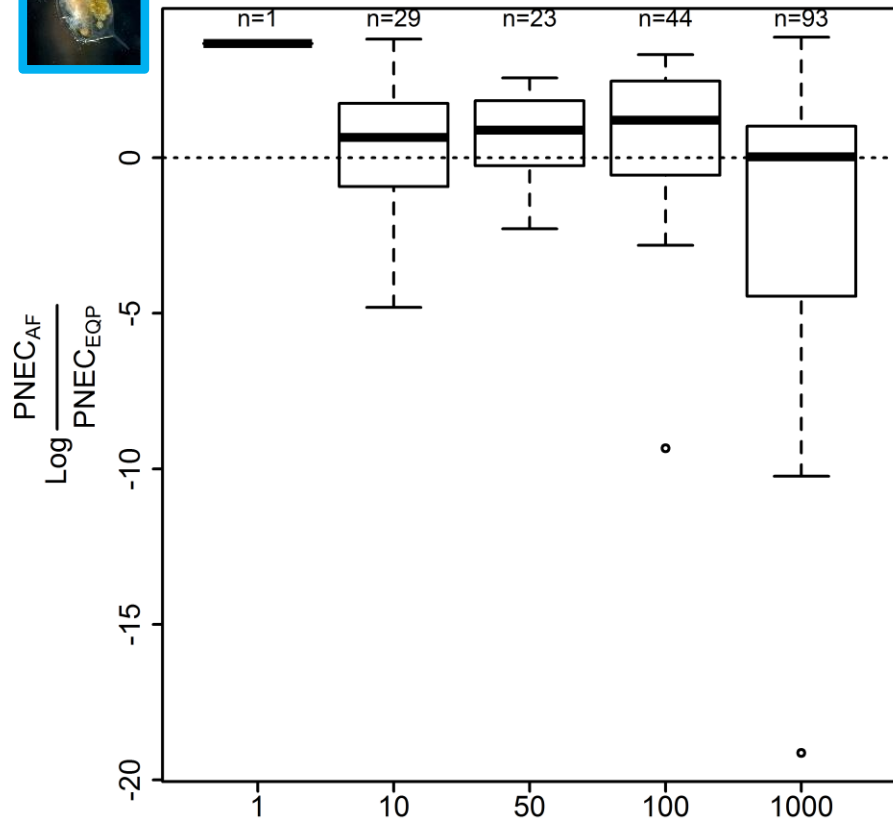
Risk assessment (IV) – PNEC/Risk



Risk assessment (IV) – PNEC/Risk

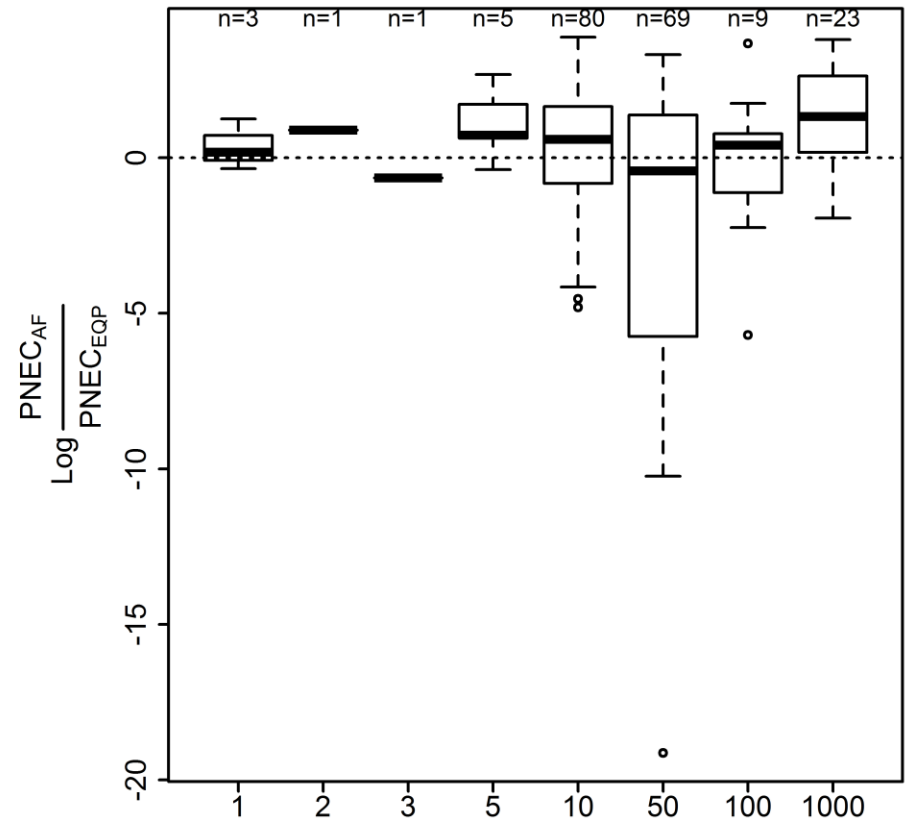


Soil



PNEC soil AF

Soil



PNEC freshwater AF



Conclusion & Suggestions

- Role of MOA unclear
- There is a clear link with K_{oc} (and VP?)
 - The hazard to soil invertebrates relative to aquatic invertebrates seems to increase with rising K_{oc} (and declining VP?)
 - This last observation affects soil RAs based only on aquatic data, warranting caution when using EqP on substances with increasing K_{oc}
 - The data suggest that
 - differences in partitioning and/or equilibrium state between aquatic and soil ecotoxicity studies could be the underlying cause
 - The role of contaminant uptake via food is unclear

THANKS



- Stewards and members of the TF

Christian Bögi (BASF), Anne-Lise Mandrillon (Solvay), Christopher Hughes (Shell), Mark Lampi (ExxonMobil), Malyka Galay Burgos (ECETOC), David Huggett (Syngenta), Emily Rogevich Garman (Nipera), Hugo Waeterschoot (Eurometaux), Chris Money (Cynara Consulting Ltd), Johannes Tolls (Henkel)

- ECETOC office for support

Christine Yannakas, Diana Buksa, Ian Cummings

- ECHA

- **For your attention!**